# metal-organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 298 KMean  $\sigma$ (C–C) = 0.007 Å R factor = 0.047 wR factor = 0.115 Data-to-parameter ratio = 13.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# A trinuclear manganese(II) complex of 2-acetyl-5-methoxyphenol

In the title centrosymmetric compound, hexakis( $\mu$ -2-acetyl-5methoxyphenolato)trimanganese(II), [Mn<sub>3</sub>(C<sub>9</sub>H<sub>6</sub>O<sub>3</sub>)<sub>6</sub>], three Mn<sup>II</sup> atoms are linearly linked by deprotonated hydroxyl groups of paeonol anions. One Mn atom is located on an inversion center and is coordinated by six hydroxy O atoms, and the Mn atom located on a general position is coordinated by three hydroxy O and three acetyl O atoms. Within the trinuclear molecule, there is a short Mn···Mn separation of 3.0464 (8) Å.

#### Comment

2-Acetyl-5-methoxyphenol (common name paeonol) is an effective component of many traditional Chinese medicines. Many new metallic complexes of paeonol have been prepared and characterized; however, structural studies of them are rarely reported (Sillanpaa, 1991; Demertzi *et al.*, 2000). Linear trinuclear manganese complexes bridged by acetate anions have been reported previously (Tangoulis *et al.* 1996; Baldwin *et al.* 1995). Recently, we prepared the title Mn<sup>II</sup> complex, (I), with paeonol ligands and present here its structure.



The trinuclear molecular structure of (I) is shown in Fig. 1. Three  $Mn^{II}$  atoms are linked linearly by deprotonated hydroxyl groups of paeonol anions. Atom Mn1 is located on an inversion center and coordinated by six hydroxy O atoms

© 2006 International Union of Crystallography All rights reserved Received 10 July 2006 Accepted 28 July 2006 with a distorted octahedral geometry. Atom Mn2 is located on a general position and coordinated by three hydroxy O and three acetyl O atoms, with a distorted octahedral geometry (Table 1). Within the trinuclear molecule, there is a short  $Mn \cdots Mn$  separation of 3.0464 (8) Å.

Weak  $C-H \cdots O$  interactions are observed in the crystal structure (Table 2).

### **Experimental**

A methanol solution (20 ml) of manganese(II) acetate dihydrate (0.184 g, 1.0 mmol) was mixed with a methanol solution (15 ml) of paeonol (0.166 g, 1 mmol). After stirring for 3 h at 320 K, the precipitate was filtered off. Single crystals of (I) were obtained by slow evaporation of the filtrate after 9 d.

Z = 2

 $D_x = 1.456 \text{ Mg m}^{-3}$ 

 $0.19 \times 0.18 \times 0.12 \ \text{mm}$ 

13782 measured reflections

4647 independent reflections 2690 reflections with  $I > 2\sigma(I)$ 

Mo  $K\alpha$  radiation

 $\mu = 0.78 \text{ mm}^{-1}$ 

T = 298 (2) K

Block, yellow

 $R_{\rm int} = 0.052$ 

 $\theta_{\rm max} = 25.0^{\circ}$ 

Crystal data

$$\begin{split} & [\mathrm{Mn}_3(\mathrm{C_9H_6O_3})_6] \\ & M_r = 1155.79 \\ & \mathrm{Monoclinic}, \ P2_1/n \\ & a = 11.967 \ (2) \ \mathrm{\AA} \\ & b = 19.883 \ (3) \ \mathrm{\AA} \\ & c = 12.304 \ (2) \ \mathrm{\r{A}} \\ & \beta = 115.740 \ (2)^\circ \\ & V = 2637.1 \ (7) \ \mathrm{\r{A}}^3 \end{split}$$

#### Data collection

Bruker SMART CCD area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 2002)  $T_{\min} = 0.866, T_{\max} = 0.912$ 

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.032P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.047$	+ 2.7402P]
$wR(F^2) = 0.115$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.01	$(\Delta/\sigma)_{\rm max} = 0.001$
4647 reflections	$\Delta \rho_{\rm max} = 0.53 \ {\rm e} \ {\rm \AA}^{-3}$
346 parameters	$\Delta \rho_{\rm min} = -0.28 \text{ e} \text{ \AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected bond lengths (Å).

Mn1-O1	2.178 (3)	Mn2-O4	2.172 (3)
Mn1-O4	2.192 (3)	Mn2-O5	2.122 (3)
Mn1-O7	2.162 (3)	Mn2-O7	2.142 (3)
Mn2-O1	2.163 (3)	Mn2-O8	2.128 (3)
Mn2-O2	2.134 (3)		

Table	2
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Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$C25-H25\cdots O5^i$	0.93	2.55	3.396 (5)	151
Symmetry code: (i) x	$+\frac{1}{2}, -y + \frac{3}{2}, z +$	- 1/2.		



Figure 1

The molecular structure of (I), shown with 30% probability displacement ellipsoids, H atoms have been omitted for clarity. Hollow unlabeled atoms are related by the symmetry operator (1 - x, 1 - y, 1 - z).

Methyl H atoms were placed in calculated positions, with C–H = 0.96 Å, and torsion angles were refined to fit the electron density,  $U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm C})$ . Aromatic H atoms were placed geometrically, with C–H = 0.93 Å, and refined in riding mode, with  $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$ .

Data collection: *SMART* (Bruker, 2003); cell refinement: *SAINT* (Bruker, 2003); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

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#### References

- Baldwin, M. J., Kampf, J. W., Kirk, M. L. & Pecoraro, V. L. (1995). Inorg. Chem. 34, 5252–5260.
- Bruker (2003). SAINT (Version 6.45A) and SMART (Version 5.059). Bruker AXS Inc., Madison, Wisconsin, USA.
- Demertzi, D. K., Kourkoumelis, N., Demertzis, M. A., Miller, J. R., Frampoton, C. S., Swearingen, J. K. & West, D. X. (2000). *Eur. J. Chem.* pp. 727– 730.
- Sheldrick, G. M. (1997a). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (2002). SADABS. Version 2.03. University of Göttingen, Germany.
- Sillanpaa, E. R. J. (1991). Polyhedron, 10, 2051-2153.
- Tangoulis, V., Malamatari, D. A., Soulti, K., Stergiou, V., Raptopoulou, C. P., Terzis, A., Kabanos, T. A. & Kessissoglou, D. P. (1996). *Inorg. Chem.* 35, 4974–4983.